

## 205. 1-Substituted 1,3-Dihydroisothianaphthen-2,2-dioxides: Preparation and Use as *ortho*-Quinodimethane Precursors in Intramolecular Cycloadditions

Preliminary Communication

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Dedicated to Professor *André Dreiding* on his 60th birthday

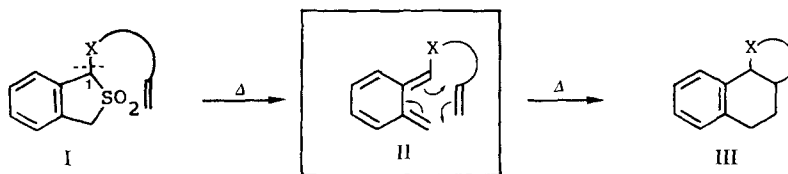
(15.VIII.79)

### Summary

1,3-Dihydroisothianaphthen-2,2-dioxide (**1**) was readily converted to the 1-substituted sulfones **3** by deprotonation and subsequent electrophilic attack (*Scheme 3* and *Table*). The appropriate 1-alkenyl- and 1-alkenoyl-sulfones **3** on heating at 213° to 240° underwent SO<sub>2</sub>-extrusion to give, *via* the non-isolated (*E*)-quinodimethanes **II** (*Scheme 1*), polycyclic products such as **4**, **6** and **7** in good yields (*Schemes 4* and *5*). On the other hand, thermolysis of the 1-alkenoyl-1-thioether sulfones **9** furnished mainly the isochromenes **10** (*Scheme 6*).

The utility of the reaction **II** → **III** (*Scheme 1*) for the general, flexible and stereoselective synthesis of polycyclic systems has been amply demonstrated by work from this and other laboratories<sup>1)</sup>. Although the transient dienes **II** are conveniently

*Scheme 1*

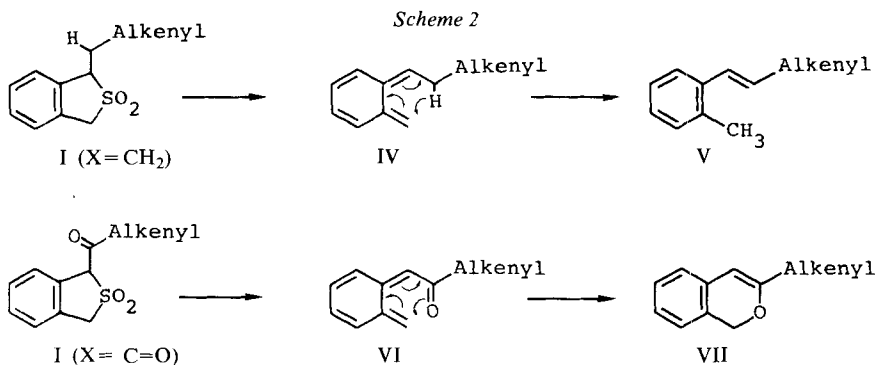


generated and converted to **III** by heating suitable 1-substituted benzocyclobutenes [1] it seemed worthwhile to explore further routes which could lead to **II**<sup>2)</sup>. Thus, in view of the postulated formation of the unstable *o*-xylylene on heating 1,3-dihydroisothianaphthen-2,2-dioxide [4], we decided to study the thermolysis of the

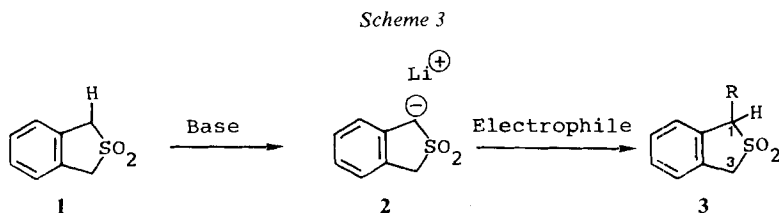
<sup>1)</sup> Review: [1].

<sup>2)</sup> See for example the preparation and thermolysis of 4-alkenyl-3-isochromanones [1] [2].

1-substituted sulfones **I**. This raised the question as to whether the sulfones **I** would eliminate  $\text{SO}_2$  to give the desired (*E*)-diene **II** or rather the (*Z*)-isomers **IV** and **VI** which would be expected to undergo 1,5-H-shift **IV**  $\rightarrow$  **V** or cyclization **VI**  $\rightarrow$  **VII**, respectively (*Scheme 2*) [1]. Despite this initial uncertainty the apparent accessibility



of **I** from the readily available sulfone **1** [4b] by formation of the C(1), X-bond justified this approach. By analogy to the recently studied alkylation of 3-isochromanones [1] [2] and of benzocyclobutyl phenyl sulfone [3] it was anticipated that deprotonation of the sulfone **1** and subsequent electrophilic substitution of the anion **2** would afford the monosubstituted sulfones **3** (*Scheme 3*). In fact, as indi-



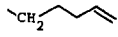

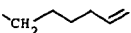
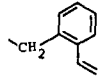
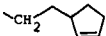


cated in the *Table*, treatment of **1** with various bases in THF under an argon atmosphere at  $-78^\circ$  for 30 min (*entries A and B*) or, more conveniently, with BuLi at  $-20^\circ$  for 15 min (*entry C*), followed by rapid addition of 1,1 mol-equiv. of methyl iodide, stirring of the mixture at  $-20^\circ$  for 30 min, aqueous work-up and crystallization of the crude product furnished the monomethylated sulfone **3a**<sup>3</sup>) in high yield.

Analogous alkylation of **1** with several, less reactive, alkenyl bromides and tosylates (*Table*) gave mainly the monosubstituted sulfones **3** in satisfactory yields after separation from minor amounts of 1,3-dialkylated products and unchanged **1**. Acylation of **1** required two mol of **2** per mol of acylating agent, and a work-up

<sup>3</sup>) IR.,  $^1\text{H-NMR}$ , and MS. are in full agreement with the assigned structure. IR. spectra: film unless otherwise specified,  $\nu_{\text{max}}$  in  $\text{cm}^{-1}$ . - UV. spectra: MeOH,  $\lambda_{\text{max}}$  in nm,  $\log \epsilon$  in parentheses. -  $^1\text{H-NMR}$ . spectra in  $\text{CDCl}_3$  at 100 MHz, internal standard tetramethylsilane ( $\delta = 0$  ppm), abbreviations: s = singlet, d = doublet, J = spin-spin coupling constant (Hz).

with acid, owing to the acidic nature of the acylsulfone products; the latter were formed in higher yields when esters were used instead of acid chlorides (see for example the preparation of **3f**, *Table, entries I and J*). Similarly, sulfenylation of the anion **2** with 0.5 mol-equiv. of the appropriate disulfide furnished the thioethers **3g**<sup>3</sup>) and **3h**<sup>3</sup>) in good yields. Furthermore, electrophilic attack of **2** by 1 mol-equiv. of 5-hexenyl-carboxaldehyde proceeded smoothly at  $-78^\circ$  to  $-20^\circ$  to give, after aqueous work-up, a stereoisomeric mixture of 1-(1-hydroxy-5-hexenyl)-1,3-dihydroisothianaphthen-2,2-dioxides<sup>3</sup>) in 79% yield. Having a practical route to various olefinic sulfones in hand we turned our attention to the anticipated thermolyses  $I \rightarrow II \rightarrow III$ .

Table. Deprotonation and Substitution of 1,3-Dihydroisothianaphthen-2,2-dioxide by the Electrophiles  $R-X: 1 \rightarrow 2 \rightarrow 3$

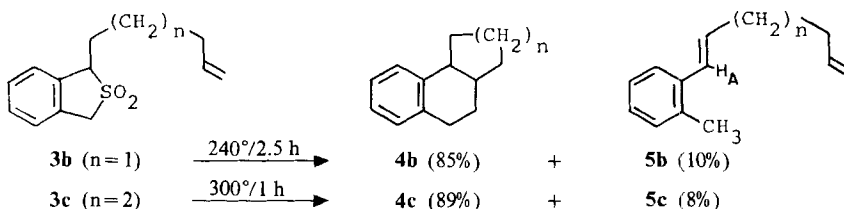
Entry	Reaction Conditions <sup>a)</sup>		X	R	Product <sup>3)</sup>	% Yield <sup>b)</sup>	M.p. (°C)	<sup>1</sup> H-NMR <sup>3)</sup> $\delta$ (H-C(1))
	1 $\rightarrow$ 2	2 $\rightarrow$ 3						
A	LiTMP/ $-78^\circ$	$-78^\circ$ to $+25^\circ$	I	CH <sub>3</sub>	<b>3a</b>	85	96-99	4.30
B	LDA/ $-78^\circ$	$-78^\circ$ /1 h	I	CH <sub>3</sub>	<b>3a</b>	92	96-99	4.30
C	BuLi/ $-20^\circ$	$-20^\circ$ /0.5 h	I	CH <sub>3</sub>	<b>3a</b>	91	96-99	4.30
D	BuLi/ $-20^\circ$ 1 eq. HMPT	$-10^\circ$ /2 h	Br		<b>3b</b>	76	oil	4.22
E	LiTMP/ $-78^\circ$ 1 eq. HMPT	$-78^\circ$ to $0^\circ$	Br		<b>3c</b>	61 (79)	oil	4.22
F	BuLi/ $-78^\circ$	$-78^\circ$ to $-10^\circ$	Br		<b>3c</b>	52 (86)	oil	4.22
G	BuLi/ $-40^\circ$	$-78^\circ$ to $-20^\circ$	Br		<b>3d</b>	53	149-150	4.43
H	BuLi/ $0^\circ$	$0^\circ$ /3 h	OTs		<b>3e</b>	45 (65)	oil	4.26
I	BuLi/ $-20^\circ$	0.5 eq. RX $-78^\circ$ /5 min	Cl		<b>3f</b>	30	oil	5.30
J	BuLi/ $-20^\circ$	0.5 eq. RX $+25^\circ$ /18 h	OEt		<b>3f</b>	95	oil	5.30
K	BuLi/ $-20^\circ$	0.5 eq. RX $-20^\circ$ /3 h	SPh	SPh	<b>3g</b>	81	77-78	5.42
L	BuLi/ $-20^\circ$	0.5 eq. RX $-20^\circ$ /3 h	SMe	SMe	<b>3h</b>	86	65-66	5.14

<sup>a)</sup> The commercially unavailable electrophiles<sup>3)</sup> were prepared by the following standard procedures: *Entry G*: see [5]; *entry H*: tosylation (1.1 mol-equiv. of TsCl/Py/ $25^\circ$ /3 h) of the known alcohol [6]; *entry I*: treatment of the 2-cyclopentene-1-acetic acid (*Aldrich*) with an excess of (COCl)<sub>2</sub>/CH<sub>2</sub>Cl<sub>2</sub>/ $25^\circ$ /18 h; *entry J*: treatment of the above acid chloride with an excess of EtOH/Py/ $25^\circ$ /18 h. The reactions were carried out as described above for *entry C* unless otherwise specified; LiTMP = lithium 2,2,6,6-tetramethylpiperide, LDA = lithium diisopropylamide, HMPT = hexamethylphosphoric triamide. The reaction mixtures obtained in the *entries I, J, K and L* were quenched with aq. 2N HCl.

<sup>b)</sup> The products **3** were obtained by crystallization, distillation and/or chromatography on SiO<sub>2</sub>; yields in parentheses are based on recovered **1**, yields cited in the *entries I, J, K and L* are based on the corresponding electrophile.

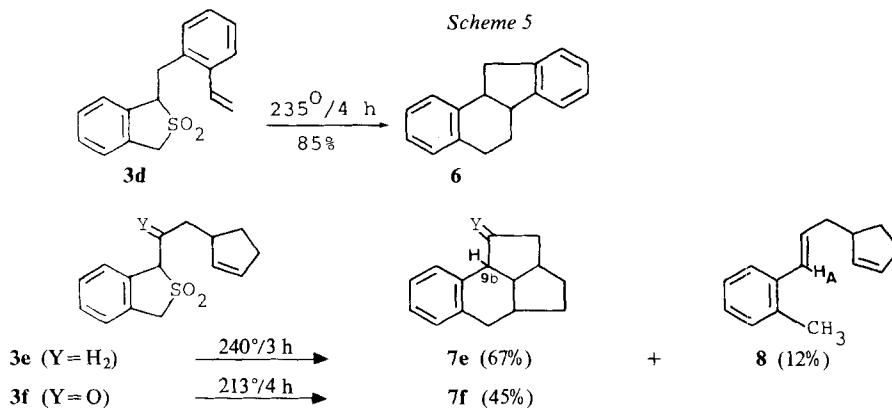
On heating the pentenyl-sulfone **3b** in diethylphthalate<sup>4)</sup> at 240° the extrusion of SO<sub>2</sub> was completed within 2.5 h to give after chromatography the desired adducts **4b**<sup>3)</sup> <sup>5)</sup> in 85% yield (Scheme 4). The minor formation of the styrene **5b**<sup>3)</sup> <sup>6)</sup> (10% yield) indeed reflects the intermediacy of some (Z)-quinodimethane IV. Similar yields of **4c**<sup>3)</sup> <sup>5)</sup> <sup>7)</sup> and **5c**<sup>3)</sup> <sup>6)</sup> were obtained on the analogous pyrolysis<sup>4)</sup> of the homologous hexenyl-sulfone **3c**.

Scheme 4



Tetracyclic systems may be readily formed when the bridge or the olefinic bond of I are part of a ring as illustrated by the efficient conversions<sup>4)</sup> **3d** → **6**<sup>3)</sup> <sup>5)</sup> and **3e** → **7e**<sup>3)</sup> <sup>5)</sup> (Scheme 5); the latter reaction again furnished the styrene **8**<sup>3)</sup> <sup>6)</sup> in low yield (12%). Thermolysis of the ketone **3f** in refluxing 1,2,4-trichlorobenzene gave the crystalline polyfused cyclopentanone **7f**<sup>3)</sup> <sup>8)</sup> as a single stereoisomer; the corresponding isochromene VII was isolated from the reaction mixture in 41% yield.

Scheme 5



4) 10% solutions of the appropriate alkenyl sulfones in diethylphthalate were heated under an argon atmosphere.

5) The <sup>13</sup>C-NMR. spectrum agrees with the assigned structure and indicates the presence of two stereoisomers.

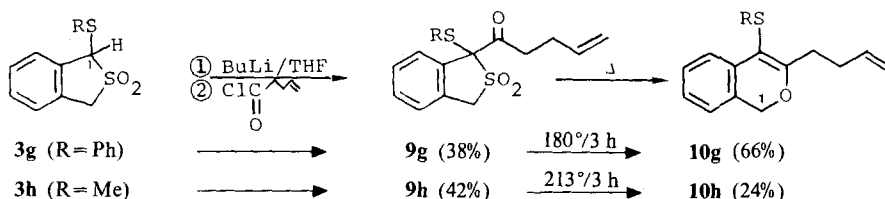
6) The styrenes **5** and **8** show in the <sup>1</sup>H-NMR. spectrum the *o*-CH<sub>3</sub> signal between δ=2.34 and 2.38, and the H<sub>A</sub>-doublet (*J*=16) between δ=6.62 and 6.66. Their UV. spectra: 209 (4.32 to 4.35), 250 (4.10 to 4.14) are typical for *o*-substituted styrenes [7].

7) Comparison of the <sup>13</sup>C-NMR. spectrum of the thermolysis product **4c** with those of independently prepared samples of *cis*- and *trans*-**4c** [8] showed the *trans*-isomer to predominate.

8) M.p. 104–107° (MeOH); IR. (CCl<sub>4</sub>): 1745; the <sup>13</sup>C-NMR. spectrum confirms the stereoisomeric purity of **7f** which on the basis of the H–C(9b) signal in its <sup>1</sup>H-NMR. spectrum at δ=3.61 (*d* × *d*, *J*=10 and 1) was assigned the all-*cis*-configuration.

On the other hand, the ketothioethers **9**<sup>3</sup>), prepared from **3g** and **3h**<sup>9</sup>) on heating in refluxing *o*-dichlorobenzene or 1,2,4-trichlorobenzene, respectively, furnished exclusively the isochromenes **10**<sup>3</sup>)<sup>10</sup>) (Scheme 6), apparently *via* quinodimethane intermediates containing a *cis*-orientated carbonyl group (in analogy to the presumed cyclization VI  $\rightarrow$  VII) (Scheme 2)<sup>11</sup>).

Scheme 6



Further work is in progress concerning the functionalization **1**  $\rightarrow$  **3**, in particular involving aryl-substituted dihydroisothianaphthendioxides, and to apply this sequence to the synthesis of natural products. We were pleased to hear very recently that 3-desoxyestrone has been synthesized independently *via* an analogous approach from **1**<sup>12</sup>).

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## REFERENCES

- [1] *W. Oppolzer*, *Synthesis* 1978, 793.
- [2] *W. Oppolzer & B. Delpech*, unpublished work.
- [3] *B.D. Gowland & T. Durst*, *Canad. J. Chemistry* 57, 1462 (1979).
- [4] a) *M.P. Cava & A.A. Deana*, *J. Amer. chem. Soc.* 81, 4266 (1959); b) *J.A. Oliver & P.A. Ongley*, *Chemistry & Ind.* 1965, 1024.
- [5] *J. von Braun*, *Ber. deutsch. chem. Ges.* 50, 45 (1917).
- [6] *D.L. Garin*, *J. org. Chemistry* 34, 2355 (1969).
- [7] *E.A. Braude & F. Sondheimer*, *J. chem. Soc.* 1955, 3773.
- [8] *R.L. Funk & K.P.C. Vollhardt*, *J. Amer. chem. Soc.* 98, 6755 (1976).
- [9] *R. Hug, H.-J. Hansen & H. Schmid*, *Helv.* 55, 10 (1972).

<sup>9</sup>) **3g** and **3h** were treated successively with BuLi (1.05 mol-equiv.) and allylacetyl chloride (1.3 mol-equiv.) in THF at  $-78^\circ$ ; the mixture was allowed to warm up to  $+25^\circ$  and subjected to aqueous work-up.

<sup>10</sup>) **10g**. – UV.: 209 (4.42), 226 (4.25), 248 (4.13), 280 (3.99). – IR.: no C=O, 1590, 1475, 908, 755. – <sup>1</sup>H-NMR.: 2 H-C(1) at  $\delta=5.2$  (s). – **10h**. – UV.: 208 (4.21), 226 (4.06), 289 (3.93). – IR. (CCl<sub>4</sub>): no C=O, 1598, 1480, 900. – <sup>1</sup>H-NMR.: 2 H-C(1) at  $\delta=5.0$  (s).

<sup>11</sup>) See also the thermolysis of 1-acylbenzocyclobutenes [9].

<sup>12</sup>) Reported by *K.C. Nicolaou* at the 6th International Symposium 'Synthesis in Organic Chemistry', Cambridge, 24–26 July 1979.